## Facile Synthesis of Furan-3,4-diacetates

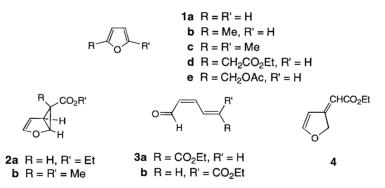
by Ernest Wenkert\* and Haripada Khatuya<sup>1</sup>)

Department of Chemistry and Biochemistry, University of California-San Diego, La Jolla, CA 92093, USA

Dedicated to the memory of Professor Paolo Ceccherelli

A procedure for bis-cyclopropanation of furans with ethyl diazoacetate or methyl  $\alpha$ -diazopropionate under dirhodium-tetraacetate catalysis is presented. Treatment of the products with ethanolic HCl furnished furan-3,4-diacetates.

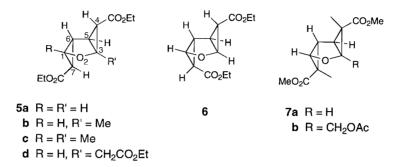
**Introduction.** – In an exhaustive study of the reaction of furan (1a) with ethyl diazoacetate (in furan as solvent), catalyzed by dirhodium tetraacetate, it was discovered that, in 66% overall yield, the reaction had furnished cyclopropafurancarboxylate 2a and furan-unravelled 6-oxohexadienoate 3a as major products and substances 3b and 4 as minor compounds of the product mixture [1]. This observation and the one showing iodine-induced product isomerization without prior product separation affording solely one compound in high yield (*e.g.*,  $1 \rightarrow 3a$ ) [1] has led to a broad investigation of this two-step, 'one-pot' procedure for the preparation of (1E,3E)-1,4-diacylbuta-1,3-dienes [1-4].



**Results and Discussion.** – It now became of interest to pursue reactions of furans with  $\alpha$ -diazocarbonyl compounds without furan-ring opening, *i.e.*, preparation of materials of type **2**, or preferably, bis-cyclopropanated substances. This required the use of excess of diazo compound and, hence, a solvent other than the furans. CH<sub>2</sub>Cl<sub>2</sub> was chosen for all the cyclopropanations, and the study was initiated with ethyl cyclopropa-

<sup>&</sup>lt;sup>1</sup>) Present address: *The R.W. Johnson PRI*, 3535 General Atomics Court, Suite 100, San Diego, CA 92121, USA.

furancarboxylate **2a**.  $[Rh_2(OAc)_4]$ -Catalyzed interaction of the latter in CH<sub>2</sub>Cl<sub>2</sub> with ethyl diazoacetate afforded diester **5a** (76%). When the same reaction was carried out on furan (**1a**), diesters **5a** (35%) and **6** (14%) were obtained.



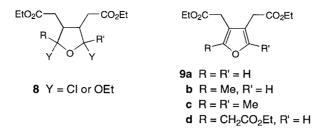
Introduction of Me groups on the furans or the diazo compounds had no ill effect on the reactions. Thus, for example, the involvement of furan (1a) with methyl  $\alpha$ -diazopropionate [5] yielded diester 7a (69%)<sup>2</sup>), and the interactions of 2-methylfuran (1b) as well as 2,5-dimethylfuran (1c) with ethyl diazoacetate gave diesters 5b (28%) (and its isomer(s), 24%) and 5c (80%), respectively<sup>3</sup>).

With dicyclopropa-furans 5-7 in hand, the time had arrived for an examination of the ease of their ring opening. To avoid the production of (1E,3E)-1,4-diacylbuta-1,3-diene equivalents (*cf.* 3) [1], reaction conditions had to be found which would mask intermediates of the enol-ether type and, thus, prevent their participation in the unravelling process. After much experimentation, refluxing ethanolic HCl was chosen for the task. Treatment of diesters 5a, 6, 5b, and 5c in this manner furnished furan-diacetates 9a (58%), 9a (46%), 9b (39%), and 9c (37%), respectively (presumably *via* intermediates of the type 8)<sup>4</sup>)<sup>5</sup>).

<sup>&</sup>lt;sup>2</sup>) The monocyclopropanated compound **2b** could be obtained (51%) from the non-polar portion of the products of a Rh-catalyzed decomposition of methyl  $\alpha$ -diazopropionate in furan. Colorless, sweet-smelling oil. <sup>1</sup>H-NMR: 0.93 (*s*, Me); 2.84 (*dd*, *J* = 3, 6, H–C(4a)); 3.68 (*s*, MeO); 4.74 (*d*, *J* = 6, H–C(1a)); 5.26 (*t*, *J* = 2.6, H–C(4)); 6.45 (*d*, *J* = 2.4, H–C(3)). <sup>13</sup>C-NMR: 5.0 (Me); 18.5 (C(6)); 36.9 (C(5)); 51.7 (MeO); 70.4 (C(1)); 102.1 (C(4)); 148.4 (C(3)); 174.8 (C=O).

<sup>&</sup>lt;sup>3</sup>) Bis-cyclopropanations could be executed also on furans with side chains containing functional groups (*e.g.*, furan-2-acetate **1d** [6]): **1d**  $\rightarrow$  **5d** (59%). Colorless oil. <sup>1</sup>H-NMR: 1.2–1.3 (*m*, 3 Me); 1.96 (*d*, *J*=4, CHCOOEt); 2.29 (*d*, *J*=3, CHOOEt); 2.46 (*dd*, *J*=3, 4, H–C(5), H–C(6)); 2.80 (*d*, *J*=17, 1 H, CH<sub>2</sub>OOEt); 2.92 (*d*, *J*=17, 1 H, CH<sub>2</sub>OOEt); 3.84 (*d*, *J*=6, H–C(1)); 4.0–4.2 (*m*, 3 CH<sub>3</sub>CH<sub>2</sub>). <sup>13</sup>C-NMR: 13.8 (Me); 13.9 (2 Me); 26.4 (CH<sub>2</sub>COOEt); 31.4 (CHCOOEt); 31.6 (CHCOOEt); 32.4 (C(5) or C(6)); 32.6 (C(5) or C(6)); 60.3 (MeCH<sub>2</sub>); 60.5 (MeCH<sub>2</sub>); 60.6 (MeCH<sub>2</sub>); 62.8 (C(1)); 70.2 (C(3)); 169.7 (CH<sub>2</sub>COOEt); 170.0 (CO); 170.1 (CO). Similarly, **1e**  $\rightarrow$  **7b**. Colorless oil. <sup>1</sup>H-NMR: 1.41 (*s*, Me); 1.54 (*s*, Me); 2.04 (*s*, MeCO); 2.28 (*d*, *J*=6, H–C(5)); 2.54 (*s*, H–C(5) or H–C(6)); 3.66 (*s*, MeO); 3.69 (*s*, MeO); 3.84 (*d*, *J*=6, H–C(1)); 4.45 (*s*, CH<sub>2</sub>). <sup>13</sup>C-NMR: 7.7 (Me); 8.9 (Me); 20.3 (*Me*CO); 27.2 (C(4) or C(7)); 30.6 (C(4) or C(7)); 31.2 (C(5) or C(6)); 32.6 (C(5) or C(6)); 52.0 (MeO); 52.2 (MeO); 62.9 (CH<sub>2</sub>); 68.2 (C(1)); 74.7 (C(3)); 170.1 (CO); 171.6 (CO); 172.3 (CO).

<sup>&</sup>lt;sup>4</sup>) Ring opening of **5d** (CF<sub>3</sub>CO<sub>2</sub>H/120°/2 h) afforded triester **9d** (57%). Colorless liquid. <sup>1</sup>H-NMR: 1.2–1.3 (*m*, 3 Me); 3.42 (*s*, CH<sub>2</sub>–C(3) or CH<sub>2</sub>–C(4)); 3.45 (*s*, CH<sub>2</sub>–C(4) or CH<sub>2</sub>–C(3)); 3.66 (*s*, CH<sub>2</sub>–C(2)); 4.1–4.3 (*m*, 3 CH<sub>2</sub>O); 7.34 (br. *s*, H–C(5)).



**Conclusion.** – A simple, two step procedure for the synthesis of furan-3,4-diacetates **9** from 3,4-unsubstituted furans has been presented.

## **Experimental Part**

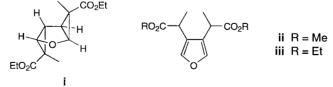
*General.* Typical reaction workup: Aq. soln. extracted with Et<sub>2</sub>O, CH<sub>2</sub>Cl<sub>2</sub>, AcOEt, the org. extract washed successively with H<sub>2</sub>O and brine, dried (Na<sub>2</sub>SO<sub>4</sub> or MgSO<sub>4</sub>), and solvent evaporated under vacuum. TLC: *EM Laboratories* precoated silica gel *60F-25* on 0.2 mm plates; column: *EM Laboratories 60–200* mesh silica gel or *Florisil* or alumina (elution with AcOEt/hexane mixtures). MPLC: *Merck Laboratory* (A, B, C) silica-gel columns, *Fluid metering Inc.* pump. M.p.: *Reichert* micro hotstage; uncorrected. IR Spectra [cm<sup>-1</sup>]: *IBM 9000* spectrophotometer. <sup>1</sup>H- and <sup>13</sup>C-NMR Spectra (CDCl<sub>3</sub> solns.): *General Electric QE-300* spectrometer. MS: *Hewlett-Packard 5890* GC/MS spectrometers.

*Diethyl* ( $1\alpha$ , $3\beta$ , $5\beta$ , $6\alpha$ )-2-*Oxatricyclo*[ $4.1.0.0^{3.5}$ ]*heptane-4a*, $7\beta$ -*dicarboxylate* (**5a**). A soln. of N<sub>2</sub>CHCO<sub>2</sub>Et (570 mg, 5.0 mmol) in dry CH<sub>2</sub>Cl<sub>2</sub> (5 ml) was added dropwise to a stirring mixture of **2a** (462 mg, 3.0 mmol) and [Rh<sub>2</sub>(OAc)<sub>4</sub>] catalyst (3 mg) in dry CH<sub>2</sub>Cl<sub>2</sub> (20 ml) over 3 h under Ar at r.t. Stirring was continued for 0.5 h and the mixture concentrated. The concentrate, as Et<sub>2</sub>O suspension, was filtered through a short *Florisil* column and the filtrate was evaporated. Silica-gel chromatography hexane/AcOEt 19:1 afforded 550 mg (76%) of **5a**. Colorless, sweet-smelling liquid. IR (neat): 1722 (C=O). <sup>1</sup>H-NMR: 1.25 (t, J = 7, 2 Me); 1.84 (d, J = 2, H–C(4), H–C(7)); 2.40 (dd, J = 4, 5, H–C(5), H–C(6)); 3.88 (d, J = 5, H–C(1), H–C(3)); 4.10 (q, J = 7, 2 CH<sub>2</sub>O). <sup>13</sup>C-NMR: 14.0 (2 Me); 27.5 (C(4), C(7)); 29.8 (C(5), C(6)); 60.6 (2 CH<sub>2</sub>O); 63.5 (C(1), C(3)); 170.2 (2 CO). HR-MS: 240.0983 (C<sub>12</sub>H<sub>16</sub>O<sub>5</sub>; calc. 240.0988).

General Cyclopropanation Procedure. A soln. of the  $\alpha$ -diazo ester (25 mmol) in CH<sub>2</sub>Cl<sub>2</sub> (10 ml) was added dropwise to a stirring mixture of [Rh<sub>2</sub>(OAc)<sub>4</sub>] (10 mg) and the furan starting material (10 mmol) in CH<sub>2</sub>Cl<sub>2</sub> (30 ml) over a 3-h period at ambient temp. The mixture was concentrated under vacuum, filtered through a short *Florisil* column, and evaporated.

Diethyl  $(1\alpha, 3\beta, 5\beta, 6\alpha)$ -2-Oxatricyclo[4.1.0.0<sup>3,5</sup>]heptane-4 $\alpha, 7\alpha$ -dicarboxylate (6). A reaction of ethyl diazoacetate (5.70 g, 50 mmol) and **1a** (1.36 g, 20 mmol) under conditions of the *General Procedure* yielded a pale-

<sup>5</sup>) Solvolysis of diester **7a** (refluxing HCl/EtOH/3 h) was very slow, leading to transesterification product **i** (colorless liquid; <sup>1</sup>H-NMR: 1.24 (*t*, *J* = 7, 2 MeCH<sub>2</sub>); 1.46 (*s*, 2 Me); 2.20 (*d*, *J* = 5, H–C(5), H–C(6)); 3.81 (*d*, *J* = 5, H–C(1), H–C(3)); 4.11 (*q*, *J* = 7, 2 CH<sub>2</sub>). <sup>13</sup>C-NMR: 7.7 (2 Me); 14.0 (2 MeCH<sub>2</sub>); 26.9 (C(4), C(3)); 30.2 (C(5), C(6)); 60.8 (2 CH<sub>2</sub>); 68.1 (C(1), C(3)); 172.4 (2 CO)), ring-opened diester **ii** (colorless liquid; <sup>1</sup>H-NMR: 1.37 (*d*, *J* = 7, 2 Me); 3.51 (*dq*, *J* = 7, 2, 2 CHCO); 3.69 (*s*, 2 MeO); 7.50 (*d*, *J* = 2, H–C(2), H–C(5))), and transesterified, ring-opened diester **iii** (colorless liquid; <sup>1</sup>H-NMR: 1.24 (*t*, *J* = 7, 2 Me); 3.62 (*dq*, *J* = 7, 2, 2 CHCO); 4.14 (*q*, *J* = 7, 2 CH<sub>2</sub>); 7.34 (*d*, *J* = 2, H–C(2), H–C(5)). <sup>13</sup>C-NMR: 14.1 (2 *Me*CH<sub>2</sub>); 18.5 (2 Me); 35.4 (2 CH); 60.8 (2 CH<sub>2</sub>O); 123.8 (C(3), C(4)); 140.1 (C(2), C(5)); 174.2 (2 CO)).



2372

yellow oil, whose MPLC (hexane/AcOEt 19:1) gave diethyl fumarate and diethyl maleate (in the earliest eluates), 1.67 g (35%) of **5a** (spectrally identical with the above sample), and 0.68 g (14%) of colorless, liquid **6**. <sup>1</sup>H-NMR: 1.24 (t, J = 7, Me of *exo*-ester); 1.30 (t, J = 7, Me of *endo*-ester); 1.55 (dd, J = 6, 9, H–C(7)); 1.81 (dd, J = 1, 4, H–C(4)); 2.28 (dd, J = 6, 9, H–C(6)); 2.38 (dd, J = 4, 6, H–C(5)); 3.87 (t, J = 6, H–C(3)); 4.05 (br. d, J = 6, H–C(1)), 4.10 (q, J = 7, *exo*-CH<sub>2</sub>O); 4.20 (q, J = 7, *endo*-CH<sub>2</sub>O). <sup>13</sup>C-NMR: 14.2 (2 Me); 24.7 (C(7)); 26.8 (C(4)); 28.2 (C(6)); 28.9 (C(5)); 60.5 (*endo*-CH<sub>2</sub>O); 60.9 (*exo*-CH<sub>2</sub>O); 67.0 (C(1), C(3)); 168.3 (*endo*-CO); 170.6 (*exo*-CO). HR-MS: 240.0999 (C<sub>12</sub>H<sub>16</sub>O<sub>5</sub>; calc. 240.0998).

*Dimethyl* (1*a*,3*β*,5*β*,6*a*)-3,6-*Dimethyl*-2-oxatricyclo[4.10.0. $^{3.5}$ ]heptane-4*a*,7*β*-dicarboxylate (**7a**). Following the *General Cyclopropanation Procedure*, the combination of methyl *a*-diazopropionate (3.98 g, 35 mmol) and **1a** (1.03 g, 15 mmol) furnished 2.51 g (69%) of colorless, crystalline **7a**. M.p. 98°. <sup>1</sup>H-NMR: 1.46 (*s*, 2 Me); 2.21 (*d*, *J* = 6, H–C(5), H–C(6)); 3.65 (*s*, 2 MeO); 3.82 (*d*, *J* = 6, H–C(3), H–C(1)). <sup>13</sup>C-NMR: 7.7 (2 Me); 26.9 (C(4), C(7)); 30.2 (C(5), C(6)); 51.8 (2 MeO); 68.1 (C(3), C(1)); 172.6 (2 CO). HR-MS: 240.0992 (C<sub>12</sub>H<sub>16</sub>O<sub>5</sub>; calc. 240.0998).

*Diethyl* ( $1\alpha$ , $3\beta$ , $5\beta$ , $6\alpha$ )-3-*Methyl*-2-*oxatricyclo*[ $4.10.0^{3.5}$ ]*heptane*- $4\alpha$ , $7\beta$ -*dicarboxylate* (**5b**). A reaction of ethyl diazoacetate (7.12 g, 62 mmol) and 2-*methylfuran* (**1b**) (2.05 g, 25 mmol) as above yielded an oil, whose MPLC (hexane/AcOEt 19:1) afforded diethyl fumarate, diethyl maleate, and 1.78 g (28%) of colorless, liquid **5b**. IR (neat): 1754 (C=O), 1775. <sup>1</sup>H-NMR: 1.24 (t, J = 7,  $MeCH_2$ ); 1.26 (t, J = 7,  $MeCH_2$ ); 1.46 (s, Me-C(3)); 1.74 (d, J = 4, H-C(7)); 1.86 (d, J = 4, H-C(4)); 2.26 (d, J = 4, H-C(5)); 2.43 (dd, J = 4, 6, H-C(6)); 3.83 (d, J = 6, H-C(1)); 4.10 (q, J = 7,  $CH_2O$ ); 4.12 (q, J = 7,  $CH_2O$ ). <sup>13</sup>C-NMR: 13.9 (3 Me); 27.9 (C(7)); 31.6 (C(6)); 32.2 (C(4)); 32.8 (C(5)); 60.3 (2 CH<sub>2</sub>O); 62.6 (C(1)); 70.2 (C(3)); 169.9 (2 CO). HR-MS: 254.1149 (C<sub>13</sub>H<sub>18</sub>O<sub>5</sub>; calc. 254.1154).

A subsequent eluate led to 1.52 g (24%) of liquid isomer(s) of **5b**. HR-MS: 254.1157 ( $C_{13}H_{18}O_5$ ; calc. 254.1154)<sup>6</sup>).

*Diethyl* (1*a*,3*β*,5*β*,6*a*)-1,3-*Dimethyl*-2-oxatricyclo[4.1.0.0<sup>3.5</sup>]heptane-4*a*,7*β*-dicarboxylate (**5c**). Ethyl diazoacetate (2.85 g, 25 mmol) interacted with 2,5-dimethylfuran (**1c**) (0.96 g, 10 mmol) according to the above method, and MPLC of the crude product (hexane/AcOEt 19:1) furnished 2.15 g (80%) of colorless, liquid **5c**. <sup>1</sup>H-NMR: 1.26 (t, J = 7, 2 *Me*CH<sub>2</sub>); 1.47 (s, 2 Me); 1.77 (d, J = 4, H–C(4), H–C(7)); 2.31 (d, J = 4, H–C(5), H–C(6)); 4.13 (q, J = 7, 2CH<sub>2</sub>O). <sup>13</sup>C-NMR: 13.9 (2 *Me*CH<sub>2</sub> or 2 Me); 14.1 (2 Me or 2 *Me*CH<sub>2</sub>); 32.7 (C(4), C(7)); 34.7 (C(5), C(6)); 60.4 (2 CH<sub>2</sub>O); 69.6 (C(3), C(1)); 170.1 (2 CO). HR-MS. 268.1268 (C<sub>14</sub>H<sub>20</sub>O<sub>5</sub>; calc. 268.1310).

General Cyclopropane Ring-Opening Procedure. A soln. of the bis-cyclopropane diester (1.0 mmol) in a conc. HCl/EtOH mixture 1:1 ( $\nu/\nu$ ) (4 ml) was refluxed for 2 h and then cooled to r.t. The mixture was diluted with H<sub>2</sub>O (5 ml) and extracted three times with Et<sub>2</sub>O (20 ml each). The combined extracts were dried and then evaporated. Silica-gel chromatography and elution with hexane/AcOEt 19:1 afforded oily furan-3,4-diacetate.

*Diethyl Furan-3,4-diacetate* (9a). Application of the above procedure to 300 mg of ester 5a or 6 gave each 130 mg (43%) or colorless, liquid 9a. <sup>1</sup>H-NMR: 1.27 (t, J = 7, 2 Me); 3.46 (s, 2 CH<sub>2</sub>); 4.16 (q, J = 7, 2 CH<sub>2</sub>O); 7.34 (s, H–C(2), H–C(5)). <sup>13</sup>C-NMR: 13.9 (2 Me); 29.3 (2 CH<sub>2</sub>); 60.9 (2 CH<sub>2</sub>O); 117.5 (C(3), C(4)); 141.1 (C(2), C(5)); 170.8 (2 CO). MS: 240, 194, 167, 166, 149, 121, 111, 95. HR-MS (CI): 241.1053 (C<sub>12</sub>H<sub>16</sub>O<sub>5</sub> + H; calc. 241.1075).

*Refluxing* **5a** (280 mg) in CF<sub>3</sub>COOH (0.5 ml) for 2.5 h led to 160 mg (58%) of **9a**. When **6** (117 mg) and TsOH (65 mg) in CCl<sub>4</sub> (5 ml) were refluxed for 6 h, 53 mg (46%) of **9a** were obtained.

<sup>&</sup>lt;sup>6</sup>) In view of the presence of trace amounts of a substance derivable from solely diazoacetate in the product mixture of the bis-cyclopropanations, a reaction was executed under the conditions of the *General Cyclopropanation Procedure* but in the absence of any furan. This led to the usual fumarate and maleate esters, and 12% of the ubiquitous compounds **iv** [7] as a pale-yellow semi-solid: <sup>1</sup>H-NMR: 1.29 (*t*, *J* = 7, Me); 1.31 (*t*, *J* = 7, Me); 1.35 (*t*, *J* = 7, Me); 4.23 (*q J* = 7, CH<sub>2</sub>); 4.25 (*q*, *J* = 7, CH<sub>2</sub>); 4.32 (*q*, *J* = 7, CH<sub>2</sub>); 4.40 (*d*, *J* = 6, H–C(4)); 4.77 (*d*, *J* = 6, H–C(5)); 6.99 (br. *s*, NH). <sup>13</sup>C-NMR: 13.8, 13.9, 14.0 (3 Me); 52.2 (C(4)); 61.3, 62.0, 62.3 (3 CH<sub>2</sub>); 66.1 (C(5)); 139.6 (C=N); 161.1 (conjugated CO); 168.9, 169.6 (2 C=O).



iv

*Diethyl 2-Methylfuran-3,4-diacetate* (9b). Following the *General Procedure* for cyclopropane opening on 56 mg of 5b led to 22 mg (39%) of colorless, liquid 9b. <sup>1</sup>H-NMR: 1.27 (m, 2 MeCH<sub>2</sub>); 2.24 (s, Me); 3.36 (s, C(3)–CH<sub>2</sub>); 3.43 (s, C(4)–CH<sub>2</sub>); 4.16 (m, 2 CH<sub>2</sub>O); 7.24 (br. s, H–C(5)). MS: 254, 208, 181, 180, 163, 135, 125, 109.

*Diethyl* 2,5-*Dimethylfuran-3,4-diacetate* (**9c**). The same reaction on 200 mg of **5c** resulted in the formation of 79 mg (37%) of colorless, liquid **9c**. <sup>1</sup>H-NMR: 1.26 (t, J = 7, 2  $MeCH_2$ ); 2.19 (s, 2 Me); 3.34 (s, 2 CH<sub>2</sub>); 4.12 (q, J = 7, 2 CH<sub>2</sub>O). <sup>13</sup>C-NMR: 11.4 (2 Me); 14.0 (2  $MeCH_2$ ); 29.7 (2 CH<sub>2</sub>); 60.6 (2 CH<sub>2</sub>O); 112.4 (C(3), C(4)); 146.8 (C(2), C(5)); 171.3 (2 CO). MS: 268, 222, 195, 194, 177, 149, 139, 123.

## REFERENCES

- [1] E. Wenkert, M. Guo, R. Lavilla, B. Porter, K. Ramachandran, J.-J. Sheu, J. Org. Chem. 1990, 55, 6203.
- [2] E. Wenkert, M. Guo, F. Pizzo, K. Ramachandran, Helv. Chim. Acta 1987, 70, 1429.
- [3] E. Wenkert, R. Decorzant, F. Näf, Helv. Chim. Acta 1989, 72, 756.
- [4] P. Ceccherelli, M. Curini, M. C. Marcotullio, O. Rosati, E. Wenkert, J. Org. Chem. 1994, 59, 2882.
- [5] J. B. Hendrickson, W. A. Wolf, J. Org. Chem. 1968, 33, 5540; M. B. Sohn, M. Jones, Jr., M. E. Hendrick, R. R. Rando, W. von Doering, *Tetrahedron Lett.* 1972, 53.
- [6] E. Baciocchi, E. Muraglia, G. Sleiter, J. Org. Chem. 1992, 57, 6817.
- [7] O. Silberred, C. S. Roy, J. Chem. Soc. 1906, 89, 179.

Received September 8, 1998